



TABLE 1. Characteristics of the Synthesized Compounds

Compound	Molecular formula	T <sub>mp</sub> , °C*	R <sub>f</sub> * <sup>2</sup>	PMR Spectra, δ, ppm* <sup>3</sup>				Yield, %* <sup>4</sup>
				<i>t</i> -Bu (18H, s)	OH (1H, s)	H <sub>arom</sub> (2H, s)	other protons	
IX	C <sub>19</sub> H <sub>21</sub> Cl <sub>6</sub> N <sub>3</sub> O	208...210	0,64(a)	1,68	4,88	7,20	—	27 (45)
X	C <sub>20</sub> H <sub>23</sub> Cl <sub>6</sub> N <sub>3</sub> O	133...134,5	0,58(6)	1,56	4,92	7,32	3,65 (2H, c, CH <sub>2</sub> )	75 (12)
XI	C <sub>21</sub> H <sub>25</sub> Cl <sub>6</sub> N <sub>3</sub> O	Oil ( <i>n</i> <sub>D</sub> <sup>20</sup> <sub>1</sub> ,4634)	0,49(6)	1,60	5,10	7,18	4,02...4,14 (4H, m, CH <sub>2</sub> CH <sub>2</sub> )	70 (15)
XII	C <sub>20</sub> H <sub>23</sub> Cl <sub>6</sub> N <sub>3</sub> OS	157...158	0,74(a)	1,70	4,95	7,34	3,92 (2H, c, CH <sub>2</sub> )	72 (15)
XIII	C <sub>21</sub> H <sub>25</sub> Cl <sub>6</sub> N <sub>3</sub> OS	Oil ( <i>n</i> <sub>D</sub> <sup>20</sup> <sub>1</sub> ,4805)	0,52(a)	1,52	5,02	7,24	4,08...4,16 (4H, m, CH <sub>2</sub> CH <sub>2</sub> )	76 (13)
XIV	C <sub>19</sub> H <sub>21</sub> Cl <sub>6</sub> N <sub>3</sub> OS	162...163,5	0,40(6)	1,50	4,86	7,27	—	92
XV	C <sub>20</sub> H <sub>23</sub> Cl <sub>6</sub> N <sub>3</sub> OS	108...109	0,54(6)	1,65	4,98	7,15	3,90 (2H, c, CH <sub>2</sub> )	94

\*Compounds recrystallized: IX from aqueous Methyl Cellosolve; X from petroleum ether; XII from aqueous ethanol; XIV from ethanol; XV from 1:3 acetone:hexane.

\*<sup>2</sup>Solvent for the system shown in parentheses.

\*<sup>3</sup>Spectra of compounds IX and XIV were taken in DMSO-D<sub>6</sub>; of compounds X-XIII and XV, in CDCl<sub>3</sub>.

\*<sup>4</sup>Yield of by-product sym-triazene VIII show in parentheses.

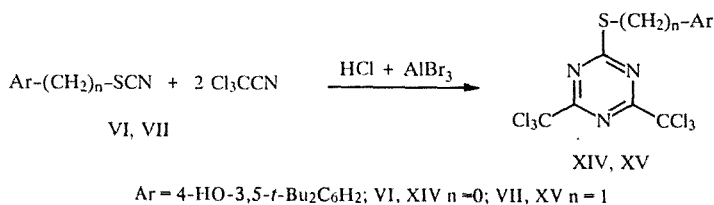
at 145-150°C. At the same time, sym-triazene VIII is formed in significant amounts (40-45% yield) along with desired sym-triazene IX.

We were not successful in increasing the yield of sym-triazene IX and lowering the yield of by-product VIII by changing the ratio of the reactants, prolonging the process, or using other catalysts (AlCl<sub>3</sub> + HCl, BF<sub>3</sub> etherate + HCl, ZnCl<sub>2</sub> + HCl, SnCl<sub>4</sub> + HCl). It should be noted that increasing the reaction temperature to 165-170°C led to the conversion of the entire reaction mixture to tar.

Presumably such a low yield of sym-triazene IX is due to the low reactivity of the cyano group in nitrile I [9].

The simultaneous cyclotrimerization of TCAN with nitriles II-V was carried out quite smoothly in the presence of gaseous HCl by the method previously developed [8] for the preparation of 6-aryl-2,4-bis(trichloromethyl)-sym-triazenes, using a TCAN:nitrile mole ratio of 1.5:1 in ether or chloroform at -5-0°C (2 h), holding the mixture at 20°C, and then heating it to 95-100°C. Under these conditions 6-substituted 2,4-bis(trichloromethyl)-sym-triazenes (X-XIII) were synthesized in 70-76% yields, although considerable amounts (12-15%) of by-product sym-triazene VIII were still isolated from the reaction mixtures. The yield of compound VIII rose to 20-25% when mixed acid catalysts (AlCl<sub>3</sub> + HCl, BF<sub>3</sub> etherate + HCl) were used in the reactions of TCAN with nitriles II-V.

It is known [10] that 2,4-bis(trichloromethyl)-6-alkyl(aryl)thio-sym-triazenes are readily formed by the simultaneous cyclotrimerization of TCAN with thiocyanates in the presence of AlBr<sub>3</sub> and HCl gas. In the present work it was shown that thiocyanates VI and VII react with TCAN (1:2 mole ratio) in the presence of AlBr<sub>3</sub> and gaseous HCl at -20-0°C to give the corresponding 2,4-bis(trichloromethyl)-6-thio-sym-triazenes (XIV, XV). Note that in the case discussed, not even a trace of by-product VIII was found in the reaction mixture.



In the IR spectra of synthesized sym-triazenes VIII-XV, absorption maxima of varying intensities were found which could be assigned to stretching ( $1530-1520$ ,  $1410-1400\text{ cm}^{-1}$ ), breathing ( $1110-1095$ ,  $1005-995\text{ cm}^{-1}$ ), out-of-plane bending ( $805-790\text{ cm}^{-1}$ ), and in-plane ( $705-685\text{ cm}^{-1}$ ) bending vibrations of the sym-triazene ring [2-5, 8, 11, 12]. Intense absorption bands in the  $780-770\text{ cm}^{-1}$  region correspond to the C-Cl stretching vibration [8]. Along with these vibrations, the spectra of compounds IX-XIII, XIV, and XV also have absorption bands due to the sterically hindered phenol group: a narrow band at  $3655-3640\text{ cm}^{-1}$ , characteristic of a shielded hydroxyl [13]; two bands of medium intensity in the  $1260-1210\text{ cm}^{-1}$  range attributed to Ar-OH stretching in shielded phenols [14], and two groups of bands in the  $885-870$  and  $830-820\text{ cm}^{-1}$  region (out-of-plane bending vibrations of a tetrasubstituted benzene ring).

In the PMR spectra (see Table 1) sym-triazenes IX-XIII, XIV, and XV the signals from the hydroxyl protons occur as singlets in the  $4.86-5.10\text{ ppm}$  range. This is characteristic of shielded phenols [13, 15]. Signals of the tert-butyl group protons are found as singlets in the  $1.50-1.70\text{ ppm}$  region. Singlets at  $7.15-7.34\text{ ppm}$  are due to the two magnetically shielded protons of the hydroxyaryl groups [2-5, 15].

## EXPERIMENTAL

The IR spectra were taken on a Bruker IFS-48 instrument in a mineral oil suspension or in KCl tablets. The PMR spectra were obtained on a Bruker WP-80SY instrument, TMS internal standard. The course of the reactions was monitored and the purity of the compounds obtained checked by means of TLC on Brokman  $\text{Al}_2\text{O}_3$  III st. act. in benzene-methanol solvent systems, 20:1 (a) and 30:1 (b), developed with iodine vapor. Melting points were determined on a Boetius microblock.

The elementary analyses of the compounds synthesized for C, H, Cl, and N agreed with the calculated values.

The nitriles of [3,5-di(tert-butyl)]-4-hydroxybenzoic (I) [16], [3,5-di(tert-butyl)]-4-hydroxyphenylacetic (II) [17],  $\beta$ -[3,5-di(tert-butyl)]-4-hydroxyphenylpropionic (III) [3], [3,5-di(tert-butyl)]-4-hydroxyphenylthioacetic (IV) [3], and  $\beta$ -[3,5-di(tert-butyl)]-4-hydroxyphenylthiopropionic (V) [3] acids as well as [3,5-di(tert-butyl)]-4-hydroxyphenylthiocyanate (VI) [18] and [3,5-di(tert-butyl)]-4-hydroxybenzylthiocyanate (VII) [3] were prepared by known methods.

**6-[3,5-Di(tert-butyl)-4-hydroxyphenyl]-2,4-bis(trichloromethyl)-sym-triazene (IX).** A current of dry HCl was passed into a stirred mixture of 2.88 g (12.5 mmoles) of nitrile I, 3.61 g (25 mmoles) of TCAN, and 0.33 g (1.25 mmoles) of  $\text{AlBr}_3$  in 30 ml of dry ether at  $-10^\circ\text{C}$  for 1 h. The reaction mixture was held for 12 h at  $20^\circ\text{C}$ , the solvent and excess HCl removed under reduced pressure, and the residue held for 12 h at  $145^\circ\text{C}$ . The dark mass that formed was washed with a 3% solution of  $\text{NaHCO}_3$ , then with water ( $2 \times 15\text{ ml}$ ), and extracted with hot acetonitrile ( $3 \times 30\text{ ml}$ ). The extract was dried with  $\text{Na}_2\text{SO}_4$ , concentrated to a volume of 10-12 ml and chromatographed on an  $\text{Al}_2\text{O}_3$  column ( $110 \times 5.0\text{ cm}$ ). The initial nitrile, 0.38 g, ( $T_{\text{mp}}$   $145-146^\circ\text{C}$ ,  $147^\circ\text{C}$  according to [16]) was eluted with a benzene-acetone mixture (30:1), then 1.75 g (27%) of sym-triazene IX with a 20:1 benzene-methanol mixture and 1.62 g (45%) of sym-triazene VIII with a 5:1 benzene-methanol mixture.  $T_{\text{mp}}$   $93-94^\circ\text{C}$  (from ethanol).  $R_f$  0.77 (6). According to [8],  $T_{\text{mp}}$  is  $94-95^\circ\text{C}$ .

**6-[3,5-Di(tert-butyl)-4-hydroxybenzyl]-2,4-bis(trichloromethyl)-sym-triazene (X).** A current of dry HCl was passed into a stirred mixture of 3.67 g (15 mmoles) of nitrile II and 3.25 g (22.5 mmoles) of TCAN in 35 ml of dry chloroform at  $-5-0^\circ\text{C}$  for 2 h. The reaction mixture was held for 12 h at  $20^\circ\text{C}$ , the solvent and excess HCl removed under reduced pressure, and the residue held for 2 h at  $95-100^\circ\text{C}$ . The reaction mixture was cooled to  $20^\circ\text{C}$ , extracted with ether ( $4 \times 50\text{ ml}$ ), and the extract washed with 3%  $\text{NaHCO}_3$  ( $2 \times 60\text{ ml}$ ), then with water, and dried with  $\text{Na}_2\text{SO}_4$ . The solvent was removed under reduced pressure, and the remaining oil chromatographed on an  $\text{Al}_2\text{O}_3$  column ( $90 \times 5.0\text{ cm}$ ). The small amount of initial nitrile II was eluted with a 5:1 benzene-hexane mixture, then sym-triazene X is washed out with a 25:1 benzene-methanol mixture, and sym-triazene VIII with a 5:1 benzene-methanol mixture.

The sym-triazenes XI-XIII were synthesized in analogous fashion. Compounds XI and XIII were purified by further chromatography on a silica gel column with a 30:1 chloroform-acetone mixture as eluent.

**6-[3,5-Di(tert-butyl)-4-hydroxyphenylthio]-2,4-bis(trichloromethyl)-sym-triazene (XIV).** A current of dry HCl was passed into a stirred mixture of 4.7 g (18 mmoles) of thiocyanate VI, 5.2 g (36 mmoles) of TCAN, and 0.25 g (1 mmole) of  $\text{AlBr}_3$  in 40 ml of dry ether at  $-20^\circ\text{C}$  for 2 h. The reaction mixture was held for 12 h at  $0^\circ\text{C}$ , then for 24 h at  $20^\circ\text{C}$ . The solvent and excess HCl were removed under reduced pressure. The residue was held for 1 h at  $110-120^\circ\text{C}$ , cooled to  $20^\circ\text{C}$ , and crystallized from ethanol with the addition of activated charcoal.

In analogous fashion, 6-[3,5-di-tert-butyl]-4-hydroxybenzylthio]-2,4-bis(trichloromethyl)-sym-triazene (XV) was synthesized from thiocyanate VII.

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